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[Solved] Chapter 5, Problem 1 - Biology Today and Tomorrow ...

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MindTap for Biology Today and Tomorrow with Physiology ...

B. pH is a measure of the number of H⁺ ions in a solution. 1. If H⁺ and OH⁻ are balanced, the solution is neutral, or pH 7. Pure water is neutral. 2. If there are more H⁺ ions, then the pH decreases or becomes acidic. 3. If there are more OH⁻ ions, then the pH increases or becomes basic. 4. Acids release H⁺ into solutions. 5. Bases accept H⁺ ions. 6.

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